(Even Semester) Session(2023-24) Class: B.Sc. (Sem:2nd)

Subject: Chemistry

Name of Assistant / Associate Professor	Period	Topics to be covered	Acade mic activiti es to be organiz ed	Topic of Assignments / Tests to be given to the students
Mr. Rajender Kumar	15 Jan 2084	Alkene: Nomenclature of alkenes, mechanisms of dehydration of alcohols and dehydrohalogenation of alkyl halides,. The Saytzeff rule, Hofmann elimination, physical properties and relative stabilities of alkenes Chemical reactions of alkenes mechanisms involved in hydrogenation, electrophilic and free radical additions	200	
	14 Jan. 40. 30 Jan 2024	Markownikoff's rule, hydroboration- oxidation, oxymercuration- reduction, ozonolysis, hydration, hydroxylation and oxidation with KMnO4 Arene and Aromaticity: Nomenclature of benzene derivatives:. Aromatic nucleus and side chain		
	0.94	Aromaticity: the Huckel rule, aromatic ions, annulenes up to 10 carbon atoms, aromatic, anti-aromatic and non-aromatic compounds Aromatic electrophilic substitution general pattern of the mechanism, mechanism of nitration, halogenation, sulphonation		Assignment on topic Aromaticity

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and Friedel- Crafts reaction. Energy profile diagrams. Activating, deactivating substituents and orientation.	
Dienes and Alkynes: Nomenclature and classification of dienes: isolated, conjugated and cumulated dienes. Structure of butadiene, Chemical reactions 1, 2 and 1, 4 additions (Electrophilic & free radical mechanism), Diels- Alder reaction, Nomenclature, structure and bonding in alkynes. Methods of formation. Chemical reactions of alkynes, acidity of alkynes. Mechanism of electrophilic and nucleophilic addition reactions, hydroboration- oxidation of alkynes	
Alkyl Halides: Nomenclature and classes of alkyl halides, methods of formation, chemical reactions. Mechanisms and stereochemistry of nucleophilic substitution reactions of alkyl halides, S N2 and SN1 reactions with energy profile diagrams. Aryl Halides: Methods of formation and reactions of aryl halides, The	
and redections of any addition—addition—and the elimination—addition mechanisms of nucleophilic aromatic substitution reactions. Relative reactivities of alkyl halides vs allyl, vinyl and aryl halides.	

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01	Kinetics - (Physical Chemistry) Rate of reaction, rate equation,	
(adly)	factors influencing the rate of a	
10.	reaction-concentration,	Text.
15 April	temperature, pressure, solvent,	Text.
	catalyst. Order of a reaction,	
2024	integrated rate expression for zero	
	order, first order, second and third	
16 April	order reaction	
+0	Half life mani d f	
70	Half life period of a	
30 Am	reaction. Methods of determination	
	of order of reaction, effect of	
2124	temperature on the rate of reaction -	
22	Arrhenius equation. Theories of	
7	reaction rate - Simple collision	
Carried Land	theory for unimolecular and	
	bimolecular collision. Transition	
	state theory of Bimolecular	
	reactions.	

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LESSON PLAN
Class: B.Sc.(Sem: 6th) ,(Even Semester) Session(2023-24)

			Subject: Cl	HEMISTRY
Name of Assistant / Associate Professor	Period	Topics to be covered	Academic activities to be organized	Topic of Assignments / Tests to be given to the students
Mr. Rajender Kumar	15 Jan. 2024.	Organometallic Chemistry: Definition, nomenclature and classification of organometallic compounds. Preparation, properties, and bonding of alkyls of Li, Al, Hg, and Sn a brief account of metalethylenic complexes, mononuclear carbonyls and the nature of bonding in metal carbonyls.		
	16 Jan. -10. 31 Jan. 2024.	Acids and Bases, HSAB Concept: Arrhenius, Bronsted – Lowry, the Lux – Flood, Solvent system and Lewis concepts of acids & bases, relative strength of acids & bases, Concept of Hard and Soft Acids & Bases.		
	1 Feb. +0. 15 Feb. 2024	Bioinorganic Chemistry: Essential and trace elements in biological processes, metalloporphyrin's with special reference to hemoglobin and myoglobin. Biological role of alkali and alkaline earth metal ions with special reference to Ca 2+		of ogeno- -mefallic composed-

16 Feb -10. 28 Feb 2024	Bioinorganic Chemistry: Nitrogen fixation Silicones and Phosphazenes: Silicones and phosphazenes as examples of inorganic polymers, nature of bonding in triphosphazenes		
01 Mark 15 Mark 2024	and antibonding molecular orbitals, qualitative		
16 Mary 10. 31 Mary 2024.	Photochemistry: Interaction of radiation with matter, difference between thermal and photochemical processes. Laws of photochemistry: Grotthus-Drapper law, Stark- Einstein law (law of photochemical equivalence)	Te.	
01 Aprol 4016 Aprol 2024	Photochemistry: Jablonski diagram depiciting various processes occurring in the excited state, qualitative description of fluorescence, phosphorescence, non-radiative processes (internal conversion, intersystem crossing), quantum yield, photosensitized reactions-energy transfer processes (simple examples).		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1

Object 101/2029

do	Organie Synthesis via Enulates Acidity of α-hydrogens, alkylation of diethyl malonate and ethyl acetoacetate. Synthesis of ethyl acetoacetate: the Claisen condensation. Keto-enol tautomerism of ethyl acetoacetate.	
10.	Heterocyclic Compounds Introduction: Molecular orbital picture and aromatic characteristics of pyrrole, furan, thiophene and pyridine. Methods of	
2024	synthesis and chemical reactions with particular emphasis on the mechanism of electrophilic substitution. Mechanism of nucleophilic substitution reactions in pyridine derivatives.	

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y	Comparison of basicity of pyridine, piperidine and pyrrole. Introduction to condensed five and six- membered heterocycles. Prepration and reactions of indole, quinoline and isoquinoline with special reference to Fisher indole synthesis, Skraup synthesis and Bischler-Napieralski synthesis. Mechanism of electrophilic substitution reactions of.	UNKI
SUBMITTE	quinoline and isoquinoline.	
	Synthetic Polymers Addition or chain- growth polymerization.	
00100	Free radical vinyl polymerization, ionic vinyl polymerization, Ziegler-Natta polymerization and vinyl polymers. Condensation or step growth	ACUE STATE
Kons	polymerization. Polyesters, polyamides, phenol formaldehyde resins. Natural and synthetic rubbers.	CHOO FIN
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LESSON PLAN

(Even Semester) Session(2023-24) Class: B.Sc.(Sem: 4th)

Name of	Period	Topics to be covered	Subject: CH	HEMISTRY.
Assistant / Associate Professor Mr.			Academie activities to be organized	Topic of Assignments / Tests to be given to the students
Rajender Kumar	15 Jan 2024	Applications of IR spectroscopy in structure elucidation of simple organic compounds		and the second s
	16 Jau +0. 31 Jau.	Amines: Structure and nomenclature of amines, physical properties. Separation of a mixture of primary, secondary and tertiary amines. Structural features affecting basicity of amines. Preparation of alkyl and aryl amines (reduction of nitro compounds, nitriles.	100	
	04 Feb. +0. 15 Feb. 2024	Amines: reductive amination of aldehydic and ketonic compounds. Gabriel phthalimide reaction, Hofmann bromamide reaction. Electrophilic aromatic substitution in aryl amines, reactions of amines with nitrous acid and questions.		Assignment or topic Infrared (IR) absorption spectroscopy
	16 tab.	Diazonium Salts: Mechanism of diazotisation, structure of benzene diazonium chloride, Replacement of diazo group by H, OH, F, Cl, Br, I, NO2 and CN groups, reduction of diazonium salts to hyrazines, coupling reaction and its synthetic application.		

4/	Nomenclature of Carboxylic acids, structure and bonding, physical properties, acidity of carboxylic acids, effects of substituents on acid strength. Preparation of carboxylic acids. Reactions of carboxylic acids. Hell-Volhard-Zelinsky reaction. Reduction of carboxylic acids. Mechanism of decarboxylation. Relative stability of acyl derivatives. Physical properties, interconversion of acid derivatives by nucleophilic acyl substitution. Mechanisms of esterification and hydrolysis (acidic and basic).	
	Carboxylic Acids & Acid Derivatives :Relative stability of acyl derivatives. Physical properties, interconversion of acid derivatives by nucleophilic acyl substitution. Mechanisms of esterification and hydrolysis (acidic and basic).	Test
	Chemistry of f-Block elements Lanthanides: Electronic structure, oxidation states, magnetic properties, complex formation, colour, ionic radii and lanthanide contraction, occurrence, separation of lanthanides, Lanthanide compounds. Actinides: General characteristics of actinides, chemistry of separation of Np, Pu and Am from uranium, Transuranic elements, comparison of properties of Lanthanides and actinides with transition elements. Reuthon O Drus problem.	

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